

## **Physical or Chemical Change?**

### **Station #1: Paper**

1. Write down your observations before and after the change.
2. Take a piece of paper and measure its mass, record
3. Tear the paper in half and observe.
4. Mass both halves together and record.
5. Put the paper back into the bucket.
6. Clean up your stations!!

## **Physical or Chemical Change?**

### **Station #2: Baking Soda and Vinegar**

1. Write down your observations before and after the change.
2. Measure a mini spoonful of baking soda and place it into one of the deeper depressions on the clear tray.
3. Using the eye dropper. Place 4-5 drops of vinegar onto the baking soda, and observe.
4. Clean up your station!!! (Rinse & dry the tray, and clean off the table top)

## **Physical or Chemical Change?**

### **Station #3: Salt and Water**

1. Write down your observations before and after the change.
2. Get a glass beaker and fill with 150 ml of water.
3. Place 5 g of salt into the beaker of water.
4. Stir the solution for about 2-3 min, and observe
5. Clean up your station. (Rinse out the cup and clean off the table top)

## **Physical or Chemical Change?**

### **Station #4: Pencil and Sharpener**

1. Write down your observations before and after the change.
2. Sharpen your own pencils or use the ones provided.
3. Open the container carefully and observe. Record.
4. Clean up your station.

## **Physical or Chemical Change?**

### **Station #5: pH Paper with Water and Vinegar**

1. Write down your observations before and after the change.
2. Get a paper towel.
3. Place two pieces of pH paper onto a piece of paper towel.
4. Drop one drop of water onto one of the pH papers, and observe.
5. Drop one drop of vinegar onto the other piece of pH paper, and observe
6. Note any color changes.
7. Look at our list of physical and chemical properties on the board to help with determining which kind of change this is.
8. Clean up your station.

## **Physical or Chemical Change?**

### **Station #6: Iron Metal**

1. Write down your observations before and after the change.
2. Observe the properties of the iron cube
3. Observe and compare the cube to the other piece of iron.
4. Clean up your station.

## **Physical or Chemical Change?**

### **Station #7: Ice Cubes**

1. Write down your observations before and after the change.
2. Get some paper towels for clean-up.
3. Place 1 ice cubes onto your hand and observe the changes.
4. Clean up your station (dry off your table top and throw away any paper towels)

## **Physical or Chemical Change?**

### **Station #8: Alka-Seltzer and Water**

5. Write down your observations before and after the change.
1. Place 50 ml of water into the glass beaker.
2. Place a small piece of Alka-Seltzer into the beaker and observe.
3. Clean up your station (rinse out the beaker and wipe down the table top)